

ATMOS 1020 Climate Change

## 5. Greenhouse Gases

Thomas Reichler, Dept. of Atmospheric Sciences, February 10, 2022

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### Outline

- What are the main greenhouse gases?
- Which ones are changed by human activity?
- Natural vs. anthropogenic greenhouse effect

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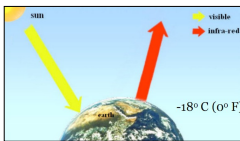
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### Primary Gases

- Our atmosphere is mostly N<sub>2</sub> (78%), O<sub>2</sub> (21%), and Ar (0.9%)
  - but these are **not greenhouse** gases
  - molecules with **1 atom** or **2 of the same atoms** aren't greenhouse gases
  - the primary gases in our atmosphere are **transparent to radiation**



If our atmosphere was only nitrogen, oxygen, and argon, this picture with no greenhouse effect would be accurate!

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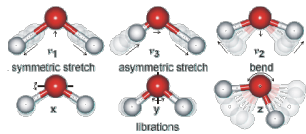
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## Major Greenhouse Gases

- Remember: greenhouse gases **block longwave radiation** from escaping directly to space; the extra downward longwave radiation from above warms the surface
- Polyatomic ( $n \geq 3$ ) molecules** are greenhouse gases
  - water vapor ( $\text{H}_2\text{O}$ )
  - carbon dioxide ( $\text{CO}_2$ )
  - methane ( $\text{CH}_4$ )
  - nitrous oxide ( $\text{N}_2\text{O}$ )
  - ozone ( $\text{O}_3$ )
  - chlorofluorocarbons (CFCs)



Vibrational, stretching, and bending modes of **water vapor** molecules mean  $\text{H}_2\text{O}$  can absorb longwave radiation

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## Greenhouse Gases

- Greenhouse gases are a tiny fraction of the atmosphere!
  - water vapor has the highest concentration:  $\sim 0.4\%$
  - carbon dioxide:  $0.04\%$  ( $\sim 400$  ppm)
  - methane:  $0.0002\%$
- These “trace gases” have a **remarkable effect**
  - e.g., ozone is less than  $0.00001\%$  of the atmosphere, but it absorbs essentially all harmful UV radiation
- Let's discuss these gases ...

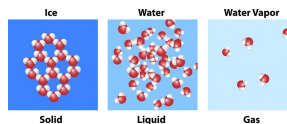


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## Water Vapor ( $\text{H}_2\text{O}$ )

- Gas form of water, invisible
  - AKA: humidity
  - not the same as steam, fog, or clouds: they are tiny water droplets or ice crystals suspended in air
- The **#1 greenhouse gas!**
  - powerful because there's a lot of it
- Not controlled by humans!
  - but observed to be increasing with global warming
  - this represents an indirect response, or a **feedback**, not a forcing (topic of a future lecture)

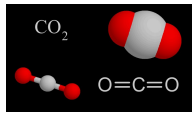


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## Carbon Dioxide (CO<sub>2</sub>)

- It's what we breathe out (respiration) and plants breathe in (photosynthesis)
- Non-toxic, invisible
- The **primary contributor** to the anthropogenic (human-caused) greenhouse effect
  - ~60% of the anthropogenic greenhouse effect so far
- Increases primarily due to
  - fossil fuel burning (80%)
  - deforestation, biomass burning, forest fires (20%)

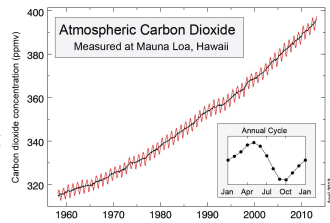


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## The "Keeling" Curve

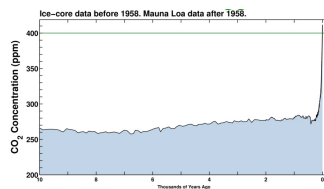
- A graph showing the variation in atmospheric CO<sub>2</sub> since 1958
- Taken at the Mauna Loa Observatory (3,400 m) in Hawaii by Charles Keeling
- Preindustrial value (1860): **280 ppm** (parts per million, = 0.028%)
- Current value (Dec 2021): **417 ppm**
- The first evidence of rapidly increasing CO<sub>2</sub> levels
- Q: What causes the seasonal wiggles?



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## Past 10,000 years

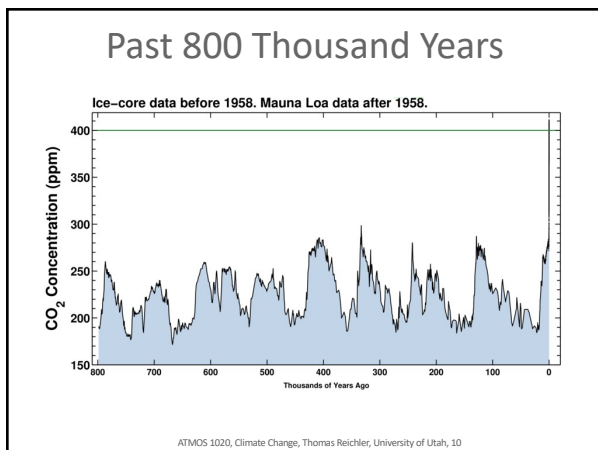
- Since 1860, CO<sub>2</sub> has risen from 280 ppm to now 417 ppm
- A 50% increase
- Entirely caused by humans
- Has a heating effect of 2 Wm<sup>-2</sup>
- Increases global temperatures by 1°C



[https://scripps.ucsd.edu/programs/keelingcurve/wp-content/plugins/sis-bluemoon/graphs/co2\\_10k.png](https://scripps.ucsd.edu/programs/keelingcurve/wp-content/plugins/sis-bluemoon/graphs/co2_10k.png)

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### The "Cyanide Cocktail"

- Some claim that 0.0137% CO<sub>2</sub> (that's the 137 ppm increase from humans) can't be bad because the number is small
- Similar claim: A man offers you a cocktail with cyanide. The man assures it is safe, since the amount of cyanide in your body after this drink would be only 0.001%!
- The effect is what counts!**
- CO<sub>2</sub> in the air could also be expressed in kg
  - it is 3200 billion tons or 3,200,000,000,000 kg
  - humans are responsible for almost 1000 billion tons, that's the 0.0137% increase

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### Carbon Dioxide In The Air

- CO<sub>2</sub> has a long life time and is fairly well mixed
- But there are tiny regional variations (here: 377 - 395 ppm) in CO<sub>2</sub> that can be detected from space or simulated by models
- The circulation transports CO<sub>2</sub> away from its source regions

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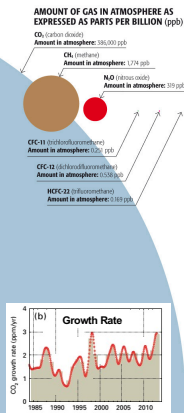
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## Carbon Dioxide (CO<sub>2</sub>)

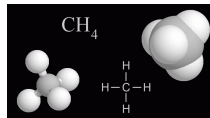
- CO<sub>2</sub> will be our main problem in the future
- it's **extremely long-lived** in the atmosphere
- 55% of what we emit quickly gets **taken up** by the ocean and land; we'll discuss this more later
- most of the rest sticks around for **over 100 years**
- some of what we emit will still be in the atmosphere over **1000 years** from now!
- Growth rate: ~2-3 ppm/yr or 0.7%



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## Methane (CH<sub>4</sub>)

- Natural gas like in stoves/heating systems
- Methane is a **much stronger greenhouse gas** on a per molecule basis than CO<sub>2</sub>
- Only 1.8 ppm though – much smaller concentration than CO<sub>2</sub>
- Natural sources
  - marshes (swamp gas) and other wetlands
- Increases anthropogenically
  - farm animals (cow burps)
  - rice farming
  - landfills
  - natural gas leakage

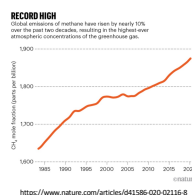


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## Methane (CH<sub>4</sub>)

- Lifetime of methane is “only” **8 years**
- This is significantly shorter than that of carbon dioxide
- CH<sub>4</sub> breaks down in the atmosphere by chemical reactions
- Concentrations have been leveling off in the early 2000s (possibly because of droughts) but increasing again since 2007



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## Global Warming Potential (GWP)

- Carbon dioxide lifetime > 100 years
- Methane lifetime = 8 years
- But per molecule, methane is a much stronger greenhouse gas
- How to put these on similar terms?
- Global warming potential (or GWP)
  - how much greenhouse effect emissions of a given gas cause over a fixed amount of time (usually 100 years)
  - measured relative to CO<sub>2</sub> (= 1)

Table ES-1: Global Warming Potentials (100-Year Time Horizon) Used in the Inventory Report

Gas	GWP
CO <sub>2</sub>	1
CH <sub>4</sub> *	21
N <sub>2</sub> O	310
HFC-23	11,700
HFC-32	650
HFC-125	2,900
HFC-134a	1,300
HFC-143a	3,800
HFC-152a	140
HFC-227ea	2,900
HFC-236fa	6,300
HFC-4310mee	1,300
CF <sub>4</sub>	6,500
C <sub>2</sub> F <sub>6</sub>	9,200
C <sub>3</sub> F <sub>8</sub>	7,000
C <sub>4</sub> F <sub>10</sub>	7,400
SF <sub>6</sub>	23,900

Source: IPCC (1996).  
\* The CH<sub>4</sub> GWP includes the direct effects and those indirect effects due to the production of tropospheric ozone and stratospheric water vapor. The indirect effect due to the production of CO<sub>2</sub> is not included.

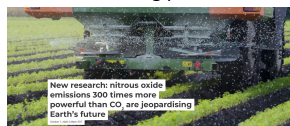
- Methane's global warming potential is **21**
  - measured over 100 years, one CH<sub>4</sub> molecule released into the atmosphere is 21 times more potent than CO<sub>2</sub> even though CH<sub>4</sub> stays around for only 8 years

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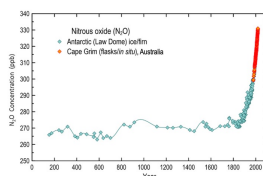
## Nitrous Oxide (N<sub>2</sub>O)

- Laughing gas (dentists use it)
- Also more potent on a per molecule basis than CO<sub>2</sub>
- Global warming potential: **310**



- Sources
  - agriculture (fertilizer)
  - chemical industry
  - deforestation

- Small concentrations of only 0.33 ppm



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## Summary So Far

- CO<sub>2</sub>, CH<sub>4</sub> and N<sub>2</sub>O all strongly increased since the industrial revolution (ca. 1750)
- Even the rate of increase is increasing!

	CO <sub>2</sub>	CH <sub>4</sub>	N <sub>2</sub> O
Global abundance in 2013 <sup>(4)</sup>	396.0±0.1 ppm	1824±2 ppb	325.9±0.1 ppb
2013 abundance relative to year 1750 <sup>a</sup>	142%	253%	121%

Source: WMO greenhouse gas bulletin 2014

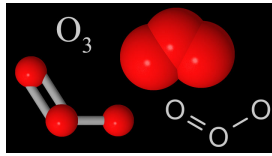
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## Ozone (O<sub>3</sub>)

Occurs in two places:

- **Stratospheric ozone**
  - good
  - natural UV protection
- **Tropospheric ozone**
  - bad
  - near the Earth's surface by air pollution (photo smog)
  - is a greenhouse gas
  - much more potent on a per molecule basis than CO<sub>2</sub>
  - very short-lived
- Global warming potential for ozone is not usually calculated – rather it's wrapped into the GWPs of the other gases that lead to its chemical creation

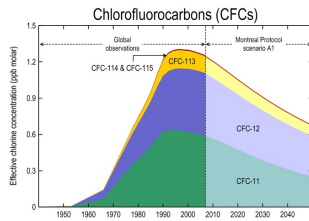


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## Chlorofluorocarbons (CFCs)

- CFCs not only destroy stratospheric ozone, they are also **very strong** greenhouse gases
- Global warming potentials of up to 11,000!
- Their reduction likely saved significant global warming in addition to the ozone layer!
- Some replacements for CFCs (called HFCs) are strong greenhouse gases too



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## The Natural Greenhouse Effect

- The greenhouse effect is a **natural phenomenon**
- We need it to keep our planet warm (0°F → 60°F)
- Contributions to the natural greenhouse effect:
  - H<sub>2</sub>O (water vapor): 60%
  - CO<sub>2</sub> (natural carbon dioxide only): 26%
  - all others: 14%

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## Planets and Atmospheres

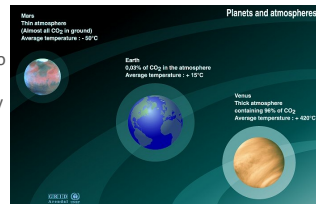
- A planet's climate is decided by its
  - mass to create gravity and to keep its atmosphere
  - sun distance for heat energy
  - atmospheric composition for greenhouse effect

### Mars

- small mass
- thin atmosphere ( $\text{CO}_2$ )
- away from sun
- 50°C

### Earth

- some greenhouse gases:  $\text{CO}_2$  (0.04%),  $\text{H}_2\text{O}$  (0-4%)
- +15°C



### Venus

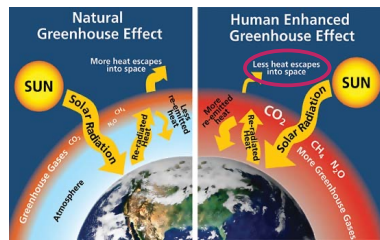
- similar mass as Earth
- thick atmosphere (96%  $\text{CO}_2$ )
- close to sun
- +420°C

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## Anthropogenic Greenhouse Effect

- Human activity has increased the concentration of certain greenhouse gases above their natural levels
- This makes it even harder for outgoing radiation to escape to space



- This **enhances the natural greenhouse effect**
- Less longwave radiation to space, less cooling**
- The **planet must warm** in order to increase its outgoing longwave radiation until new energy balance is reached

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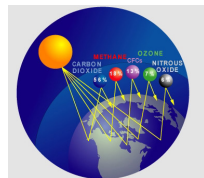
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## Anthropogenic Greenhouse Effect

- Global warming is due to an anthropogenic enhancement of the natural greenhouse effect

- Contributors to the **anthropogenic** greenhouse effect:

- carbon dioxide: 60%
- methane: 18%
- CFCs, HFCs: 13%
- tropospheric ozone: 7%
- nitrous oxide: 6%



<http://www3.int.undp.org/resources/images/ghg/ghg.html>

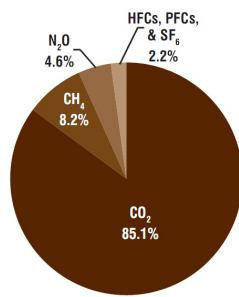
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## US Greenhouse Effect

- US contributors to the “anthropogenic” greenhouse effect based on 2008 emissions
- CO<sub>2</sub> is the big problem in the US (85% vs. 60%)
- Note how much lower the CFCs are than on the previous slide (2% vs. 13%)
  - this is b/c we basically don’t emit CFCs anymore



From US EPA 2010 report:  
[http://www.epa.gov/climatechange/emissions/downloads/10/US-GHG-Inventory-2010\\_ExecutiveSummary.pdf](http://www.epa.gov/climatechange/emissions/downloads/10/US-GHG-Inventory-2010_ExecutiveSummary.pdf)

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## Summary: Greenhouse Effect



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## Summary

- Natural greenhouse gases:
  - #1 is water vapor (60%)
  - #2 is carbon dioxide (26%)
- The greenhouse effect is a natural phenomenon, but it is enhanced by human activity on Earth
- Global warming potential: total warming caused over a fixed period of time
- Anthropogenic greenhouse effect (global warming):
  - #1 is carbon dioxide (60%)
  - methane, nitrous oxide, tropospheric ozone, CFCs are also of concern

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